## Preliminary communication

## Rigid conformation of tricyclic orthoesters of sugars

A.F. BOCHKOV, V.M. DASHUNIN, A.V. KESSENIKH, N.K. KOCHETKOV, A.D. NAUMOV, and I.V. OBRUCHNIKOV

N.D. Zelinsky Institute of Organic Chemistry, Academy of Sciences of U.S.S.R., Moscow B-334 (U.S.S.R.) (Received September 25th, 1970; accepted for publication, December 2nd, 1970)

Although a rigid conformation may be postulated for tricyclic orthoesters of sugars on the basis of general stereochemical considerations, there has previously been little experimental evidence to support this proposition. We have now synthesised the series of α-D-xylopyranose tricyclic orthoesters (1—6) and have investigated their p.m.r. spectra\*; Coxon¹ recently described the p.m.r. spectrum of the C-3 epimer of compound 5, namely, 3-O-benzoyl-α-D-ribopyranose 1,2,4-orthobenzoate (7). Our results, together with those of Coxon, provide evidence for the high conformational rigidity of the heterocyclic skeleton of compounds 1—7.

The values of the <sup>1</sup>H spin—spin coupling constants obtained for orthoesters 1–6, for solutions in deuteriochloroform or benzene, and the data of Coxon<sup>1</sup> for orthoester 7 are presented in Table I. As can be seen, the variations of the substituent at position 3 of the xylose orthoesters, the change of solvent, and inversion of the configuration at C-3 do not significantly affect the coupling constants (excluding the obvious changes in H-3 coupling due to change of configuration at C-3). The variations of dihedral angle for each pair of protons (see Table II), as calculated in accordance with the Karplus equation<sup>2</sup>, are less than 5°. Moreover, these variations in the angles are probably due to the approximate character of the Karplus equation and/or to the experimental error in the determination of coupling constants. Our results show, therefore, that the dihedral angles in these com-

<sup>\*</sup>Details of these compounds and measurements will be published later.

TABLEI

SPIN-SPIN COUPLING CONSTANTS (1) FOR THE PROTONS OF THE PYRANOID RING OF THE ORTHOESTERS 1-7

Compound	Values for J	J						1
	1,2	2,3	3,4	4,5'	4,5	5,5'	2,4	3,5
1-6a Average $b $	4.5-4.8 4.7 5.22	2.0-2.4 2.1 3.58	4.4-5.0 4.7 1.33	3.8-4.0 3.9 4.09	0.5-1.0 0.8 0.27	11.6–12.0 11.7 11.7	2.0-2.4 2.1 2.45	0.5-1.0 0.7 -0.18

<sup>d</sup>For solutions in CDCl<sub>3</sub> or in C<sub>6</sub>H<sub>6</sub> at 60 or 100 MHz, <sup>b</sup> Average values of all the data obtained for orthoesters 1-6, <sup>c</sup>The data of Coxon <sup>1</sup>, for a solution in CDCl<sub>3</sub> at 100 MHz,

TABLE II

THE DIHEDRAL ANGLES CALCULATED, ACCORDING TO THE KARPLUS EQUATION $^2$ , FROM THE J VALUES IN TABLE I FOR ORTHOESTERS  $1\!-\!7$ 

Protons	1,2	2,34	3,40	4,5'	4,5
Dihedral angles (?)	36.5-41.5	36.5-41.5 57.5-59.0	38.0-41.5 4	45,0-46.0 10	106.5111.5

<sup>a</sup>Excluding orthoester 7.

pounds are unaffected by (i) solvent, (ii) the size of the axial substituent on C-3, and (iii) inversion of configuration at C-3, indicating a system of high conformational rigidity. The rigidity of the tricyclic system in the orthoesters 1—7 makes the compounds interesting models for stereochemical and conformational investigations.

## REFERENCES

- 1 B. Coxon, Carbohyd. Res., 13 (1970) 321.
- 2 M. Karplus, J. Chem. Phys., 30 (1959) 11.

Carbohyd. Res., 16 (1971) 497-499